

Identifying the Possibilities for Superior Recovery by Pelletization of Industry Related Small and Powdery Iron Containing Waste

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Abstract: Maximal recovery of iron contents waste is an important problem in siderurgy, since its needed transformation into by-products, thus into valuable economic goods, can lead to a rational exploitation of raw material, thus, ensuring industrial needs, as well as environmental protection. Powder waste can be processed in the form of pellets or briquettes and then is used in the steelmaking processes. Thus, the powdery ferrous wastes resulting from the materials industry, from the point of view of granulation, corresponds to processing by pelletization. In line with the superior recovery of waste, our research has focused on identifying the possibilities for the pelletization of industry related the small and powdery iron containing waste that exists in very large quantities, in Hunedoara County area (Hunedoara and Calan) and beyond. This article presents the results relating to the possibilities of pelletization, of industrial small and powdery iron containing waste, from the steel industry (steel dust, mixed agglomerating and blast furnace dust, landfilled in siderurgical ponds, in vicinity of the former industrial areas of Hunedoara and Calan, Hunedoara County). The origin of the material is the minerals (bauxite residue / red mud, landfilled in vicinity of Oradea, Bihor County) or from metal surface treatment operations (coating anticorrosive dust, collected in Oradea) in the industry. The waste utilized in the laboratory experiments was processed using a series of installed equipped in the laboratories at the Faculty of Engineering in Hunedoara, University Politehnica Timisoara.

Keywords: steel dust; mixed agglomeration and blast furnace dust; bauxite residue (red mud); anti-corrosive sludge; pelletizing process; experimental pellets; compression test

1 Introduction

In the production stream of steelmaking, steel operators generate iron containing waste, continuously, in appreciable quantities, proportional to their steel production [1-7]. The main small and powdery ferrous waste, with the reported iron content are: steel dust, agglomerate dust and sludge, blast furnace dust and

sludge, converter dust and sludge, electrical steel mill electrofilter dust or a mixture of water with fine particles of material resulting from the mechanical preparation of ores or coal [7-13].

In addition to the ferrous waste resulting from the metallurgical industry, waste with ferrous content is also generated from other industries (mining and processing of minerals, energy sector, chemical industry, etc.), such as: pyritic ash from chemical industry, iron concentrate from thermal power plant ash, other waste such as lime dust, dolomite, bauxite residue (red mud), anti-corrosive sludge, galvanic sludge, etc. [1-6, 8-10].

Waste resulting from various technological processes, in particular those resulting from industrial metallurgical, mining or ore processing processes, can be processed by pelletization, briquetting and/or agglomeration, meaning that it can be used in the development of cast iron and steel production [7-26]. By processing this waste, as well as processing it into by-products, qualitatively suitable for use as raw materials or auxiliary materials in the steel industry, the areas currently occupied, can be rendered back to the surrounding landscape, thus, contributing to the greening of the general environment, as well as to the expansion of the raw material base [1-10, 13].

Pelletizing is an integral non-pressure agglomeration technique, in the mineral processing industry, that transforms various iron content materials into a premium by-products, a larger particles/agglomerates form “iron pellets”, suitable for use in an iron-making furnace, at a steel mill, such as a blast furnace or electric arc furnace [7, 9-10, 13-26]. This processing transforms the small and powdery ferrous materials into a concentrated raw material suitable for pelletizing, as opposed to being disregarded. Such minerals or raw materials, that are commonly pelletized include [1-10, 13]:

- Mined iron ore
- Other source of iron such as dust collected from blast furnaces, steel plants, converters or agglomerating sectors
- Sludge collected from mine sites such as limestone, alumina, bauxite etc.
- Sludge collected from mineral processing sectors such as red mud, galvanising sludge, coating anticorrosive sludge etc.

In order to pelletize the small and powdery ferrous materials, a binder is needed, bentonite clay being a common choice [7, 9-10, 13-26]. Various additives may also be included with the feedstock to improve performance in the steel making (graphite, lime, dolomite etc.).

In general, three primary technological phases occur in any pelletizing process:

- The mixing process (Figure 1) – that sets the stage for creating a homogeneous mixture from the small and powdery raw materials that will allow for a uniform by-product (i.e. pellets) to be created.

- The pelletizing (also named “balling”) process, in which the agglomeration operations are made using a pelletizing equipment (or balling device). At this stage, the obtained balls are referred to as “green” pellets.
- The induration process (Figure 1) – which involves a thermal treatment that heats the “green” pellets to just before their melting point, causing them to become extremely hard. In fact, the “green” pellets will be fired in order to cure into their hardened form, as strong “fired” pellets.



Figure 1

The primary technological phases in any pelletizing process

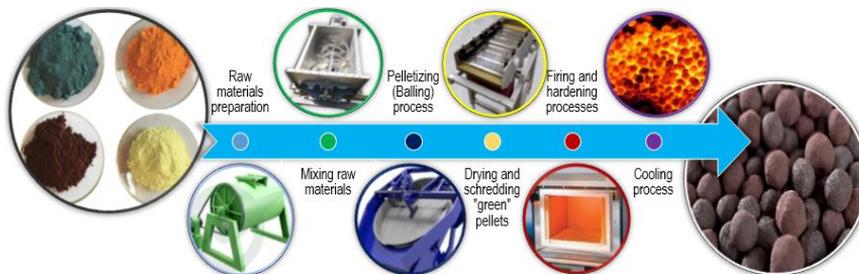


Figure 2

A complete pelletizing process



Figure 3

The by-products flow: “green” pellets – “fired” pellets – “cooled/hardened” pellets

A complete pelletizing process is typically comprised of a series of unit operations in a specific process sequence (Figure 2), in a generalized approach that can be divided into the following stages [7, 9-10, 13-26]:

- Stage of procurement and receiving raw material, the location of a pelletizing unit affecting the method of receiving raw materials such as fine and pulverous iron oxides content dust or sludge, additives and binders

- Stage of pre-processing (pre-treatment process and preparation of raw materials), suitable for pelletizing (typically involves dry grinding into finer size and screening the pulverous materials)
- Stage of pre-wetting operation that includes adding an adequate amount of water homogeneously into the dry ground material to prepare pre-wetted material suitable for balling with optimum moisture.
- Stage of proportionating and mixing the components (main ferrous content wastes, additional ferrous content wastes) with other materials called binders and additives, creating a homogeneous mixture to prepare the green balls and to achieve the required quality in final by-product (i.e. iron oxide pellet).
- Stage of pelletizing on disc pelletizer (balling process) and producing the “green” pellets from the pre-wetted mixed material prepared in the previous operation (Figure 3).
- Stage of post-processing (mainly consisting in mechanical screening operations), the “green” balls produced on the disc pelletizer being not uniform in diameter, a significant portion of the discharge (about 70%) being smaller (undersized) than target size and must be returned to disc pelletizer, after screening. If oversized balls are resulted (bigger than target size) will be returned to stage of pre-processing for dry grinding into finer size and re-mixing.
- Stage of indurating process (firing and hardening operations) the “green” pellets, establishing the binding of fine and pulverous particles at an elevated temperature, and producing the “fired” pellets (Figure 3). This is typically preceded by a drying stage, which may be carried out in the induration unit, or in a separate device, mainly in laboratory muffins furnace. After induration, the pellets can be cooled or left to cool.
- Stage of testing, often an essential part of the development of a successful pelletizing operation.
- Stage of handling of by-product (typically involves final screening for end user size requirements and storage).

2 Quality of Pellets

As a by-product, a typical iron pellet is roughly spherical in shape, measuring from 6 mm to 20 mm in diameter and having a minimum compression resistance in range of 180-200 daN/pellet or more, although some variations in these typical parameters can be specified and targeted in their preparation process [7, 9, 10, 13].

The quality of the pellets is influenced by the nature of the solid starting materials, the type and quantity of additions, the humidity of the mixture in the formation of “green” pellets and their hardening treatment [7, 10, 13]. Taking into account the demands to which pellets are subjected during transport, from the pelletization equipment to the elaboration aggregate, as well as the influence of temperature and transformations that take place in the processes of casting irons and steels manufacturing, it is necessary that they correspond to qualitative characteristics namely (chemical composition and optimal dimensions), as well as a good resistance to compression etc. [7, 10, 13].

Compression resistance is a factor of qualitative assessment of the pellets and is expressed in daN/pellet. Compression resistance is a feature that depends on the mineralogic composition of the used materials, the finesse of this material (the granulation), the addition of binder and additives, the working conditions applied and the size of the pellets [7, 10, 13]. For pellets with a diameter between 10 and 15 mm the minimum compression resistance must be in range of 180-200 daN/pellet.

Plasticity is also an important property of “green” pellets during their formation. The compression resistance of raw pellets is somewhat influenced by the plasticity of “green” pellets [7-13]. A certain degree of plasticity is required to support the growth rate of “green” pellets. If plasticity increases, the compression resistance of wet pellets decreases. The amount of moisture is required to create a certain degree of plasticity that depends on properties of particles and their distribution.

3 Materials & Methodology

The wastes from mineral processing related sectors (bauxite residue/red mud, coating anticorrosive sludge), together with the wastes from siderurgy related sector (steel dust and the mix of agglomerating/furnace dust) were subjects of the pelletizing process, in presence of graphite, used as the reducing agent, respectively bentonite and lime, used as binders [7, 9-26].



Figure 4

Main solid components: (a) steel dust / electric arc furnace dust, (b) blast furnace dust / agglomerating dust (mix), (c) bauxite residue / red mud, (d) coating sludge / anticorrosive sludge



Figure 5

Binders, additives and water: (a) bentonite clay, (b) graphite powder, (c) hydrated-lime, (d) water

For the pellets production, the ferrous waste used (Figure 4) and the manufacturing formulations are shown in Table 1, which it also shows the proportions of bentonite and lime (experiment #5 and #6), used as binders (Fig. 5), as well as the proportion of graphite used as a reducer (experiment #5 and #6). For each formulation there were produced 4 series of pellets, so a total of 24 pelletization charges.

At preparation of the experimental recipes, we have in consideration the following preliminary remarks:

- The choice of waste was made according to its chemical and mineralogic composition. The purpose was to obtain the pellets by varying the amount of very fine waste, the binder content and the amount of water used. The binder was used to increase the resistance of the pellets, in laboratory experiments being used bentonite and lime [7, 10, 13]. The results are presented in Table 1.
- With the increase in the addition of bentonite and the proportion of fine fraction, the values for compression resistance also increase. Depending on the value of the compression resistance we want to obtain and the proportion of fine fraction in the pelletization shear we choose the addition of bentonite and water. Increasing the proportion of fine fraction in the pelletization shear, requires an increase in the addition of bentonite [7, 10, 13].
- The recipes compositions and their percentual participation for the pelletizing charges in Table 1 and Figures 6-8 are presented.

The 1st experimental experimental recipe is prepared with steel dust as main component (72%), and equal percentages of red mud and anti-corrosive mud (12%). This experimental recipe is prepared without agglomerating-furnace dust, graphite and lime, but is used 4% bentonite, as is presented in Table 1 and Figure 6. The 2nd experimental recipe is prepared with agglomerating-furnace dust as main component (72%), and equal percentages of red mud and anti-corrosive mud (12%). This experimental recipe is prepared without steel dust, graphite and lime, but is used same 4% bentonite (Table 1 and Figure 6).

Table 1
Participation of solid components in experiments / recipes

Components	Participation in experiments / recipes, [%]					
	Sample #1	Sample #2	Sample #3	Sample #4	Sample #5	Sample #6
Steel dust	72	0	72	0	70	0
Agglomerating/ furnace dust	0	72	0	72	0	70
Bauxite residue (red mud)	12	12	24	0	0	24
Anti-corrosive protection mud	12	12	0	24	24	0
Graphite	0	0	0	0	1	1
Bentonite	4	4	4	4	3	3
Lime	0	0	0	0	2	2

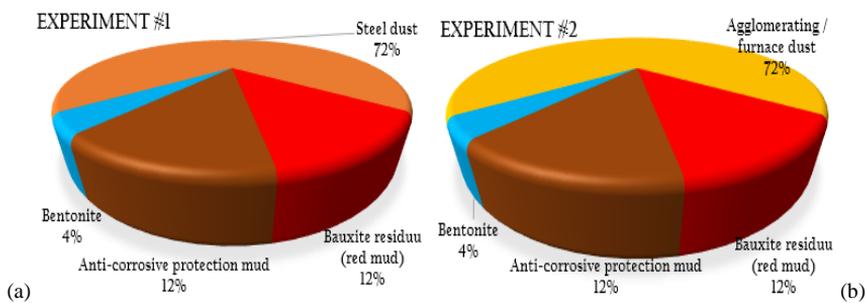


Figure 6

Participation (percentage) of solid components in Recipe #1 (a) and in Recipe #2 (b)

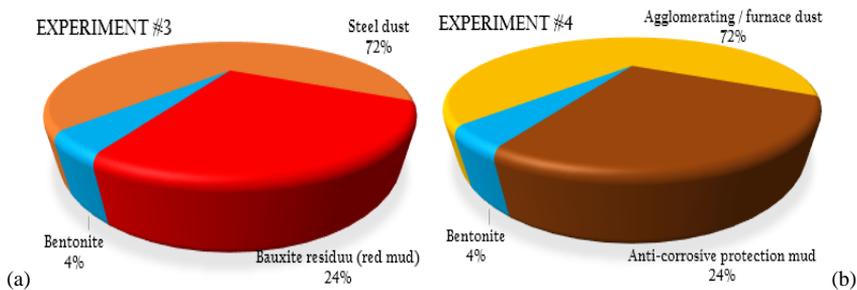


Figure 7

Participation (percentage) of solid components in Recipe #3 (a) and in Recipe #4 (b)

The 3rd experimental recipe is prepared with steel dust as main component (72%) and red mud (24%). This recipe is prepared without agglomerating–furnace dust, anti-corrosive mud, graphite and lime, but is used the same 4% bentonite (Table 1 and Figure 7). The 4th recipe is prepared with agglomerating–furnace dust as main component (72%) and anti-corrosive mud (24%). This experimental recipe is prepared without steel dust, red mud, graphite and lime, but is used the same 4% bentonite (Table 1 and Figure 7).

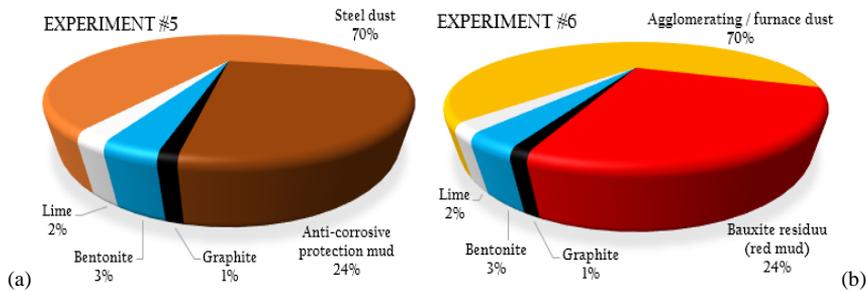


Figure 8

Participation (percentage) of solid components in Recipe #5 (a) and in Recipe #6 (b)

The 5th experimental recipe is prepared with steel dust as main component (70%) and anti-corrosive mud (24%). This recipe is prepared without agglomerating–furnace dust and red mud, but graphite (1%), lime (2%) and bentonite (3%) are used (Table 1 and Figure 8). The 6th experimental recipe is prepared with agglomerating–furnace dust as main component (70%) and red mud (24%). This recipe is prepared without steel dust and anti-corrosive mud, but graphite (1%), lime (2%) and bentonite (3%) are used (Table 1 and Figure 8).

In our research, laboratory and pilot tests were carried out on the possibilities of recovery of pulverous ferrous waste in the form of pellets [7, 10, 13]. The influence of granulation and the weight of materials of different granulometric classes in the the pelletization recipe, the proportion of binder and the addition of water on the compression resistance of “green” and “hardened” pellet was studied [7-26]. The waste envisaged in the laboratory experiments was processed, according to the technological flow shown in Figure 2.

After homogenisation (mixing the solid components), the material (40 kg material/recipe) is inserted into the experimental equipment which is a disk pelletizer that belongs to the Laboratory of Materials Processing, in Faculty of Engineering Hunedoara [7, 9-26]. The pelletizing equipment (or “balling” device) used is a disc pelletizer, for production of balls (pellets), consist of an inclined, rotating disc mounted on a stationary structure. Nucleation, compaction, size enlargement, and spheroidization of the pellets (“balling” process) take place in the course of balling and related agglomeration processes [7, 9-26].

The duration of the pelletization process is range of 10-20 min. Throughout the process, the way of wetting the composition and the formation of the pellet (“balling”) is followed. It was intended to determine the optimal additions of bentonite and water in order to obtain the most compression–resistant pellets. Depending on the humidity of the material, the water flow is adjusted. Balling has proved to be a highly versatile technique for large–scale production of particulate spheroids at relatively low capital and operational costs. Once pellets reach the desired size, they exit the pelletizing device [7, 9-26].

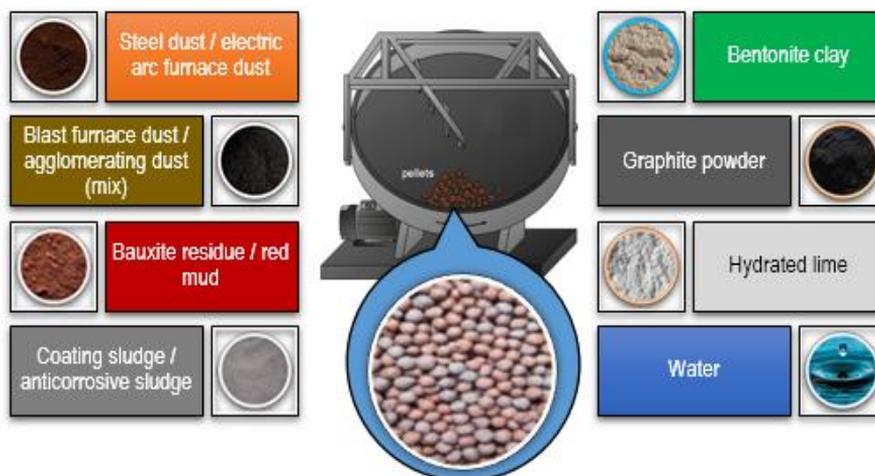


Figure 9

The pelletizing equipment and the “ingredients”

The waste used in the pelletisation process must have a fine granulation in order to form the proper pellet. For the obtained pellets, dusty wastes were used, their granulometric classes being presented in Table 2. Experimentally it was established that the size of the pellets was between 10 and 25 mm. The dimensions of 10-15 mm, a fraction representing at least 80% of the experimental recipes, were also considered to be optimal [7, 10, 13]. After completion of the pelletization operation, the obtained pellets are subjected to the granulometric sorting operation. The fraction less than 10 mm, is reintroduced into the circuit (in the pelletization stream) and the fraction greater than 10 mm will be subjected to hardening process [7, 9-26].

Table 2

Characteristics of pelletizing process and pellets

Experiment no.	Pelletizing time [min]	Humidity of green pellets [%]	Average diameter of green pellets [mm]	Granulometric classes proportion in the experimental recipes [%]					Compression resistance of pellets [daN/pellet]
				0-5 [mm]	5-10 [mm]	10-15 [mm]	15-20 [mm]	over 20 [mm]	
Sample #1	13	10.23	12.42	4	10.8	82.12	1.88	1.08	194
Sample #2	14	9.78	12.66	4.12	10.41	83.34	1.76	0.37	179.5
Sample #3	15	8.68	11.47	2.77	11.79	82.44	2.55	0.45	205
Sample #4	16	9.34	14.23	3.8	12.34	80.14	3.76	0.38	210.5
Sample #5	14	9.67	10.31	3.65	10.3	82.64	2.83	0.56	178
Sample #6	13	10.25	10.23	4.21	13.78	80.25	1.78	0.78	178.5

The hardening of the pellets are made by burning [7, 9-10, 13-26]. Burning hardening is done in electric (resistance) or flame ovens, following a proper treatment diagram (heating – maintenance – cooling), established on the basis of its own experiments (heating at 1150 °C, for 2 hours, maintenance 30 minutes and cooling in the air).

After hardening, qualitative characteristics (chemical composition, dimensional analysis) and a mechanical characteristic (compression resistance of burned pellets) were determined in our laboratories. Also, the pellets are re-examined by the granulometric sorting operation, the fraction greater than 10 mm being dispatched to the beneficiary and the fraction less than 10 mm is recirculated (reintroduced into a new pelletisation circuit) [7, 10, 13-26].

5 The Experimental Results

We start to discuss by following the graphs from the Figures 10-12, which although conclusive, are not suggestive, because it presents a cumulative analysis of all factors that compete to achieve high quality pellets, i.e. the granulation of mixed components, humidity and the duration of pelletization:

- In making the recipes it was ensured that 80% of the solid components had a grain of 10-15 mm (Figure 10, according to Table 2)
- The duration of the pelletization operation, in the average of the 4 tests/recipe, was between 13-16 minutes (Figure 11, according to Table 2), which means a good time for such processes
- Diameters obtained for pellets, in a „green” state, are between 10-15 mm (Figure 11, according to Table 2)
- The necessary humidity for the plasticity of the „green” pellets was ensured at 8-11% proportions, relative to the volume of solid quantities (Figure 11, according to Table 2)
- At first view, the compression resistance that ensure sufficient quality of the pellets are obtained with formulations #1, #3 and #4, being over 200 daN/pellet (Figure 12, according to Table 2). The other recipes provide a lower limit of quality requirements (180-200 daN/pellet). It is considered in the assessments that in most cases the pellets are intended for use in steelmaking aggregates.

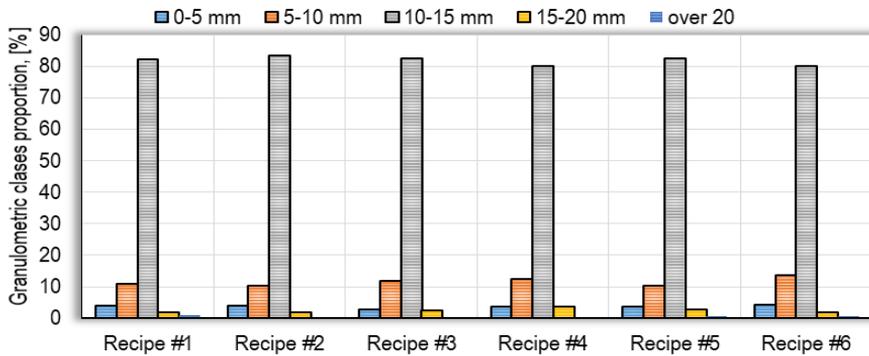


Figure 10
Granulometric classes proportion in the experimental recipes

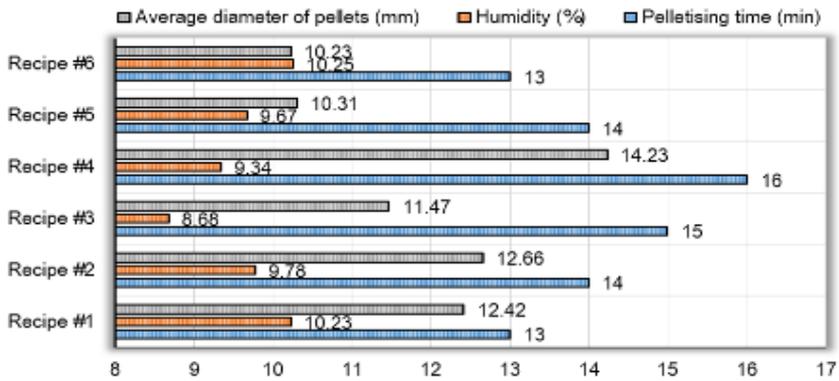


Figure 11
Average diameter of “green” pellets, humidity of “green” pellets and the pelletising time

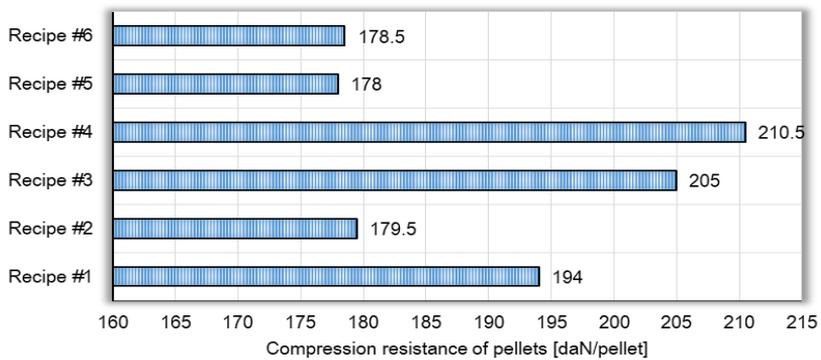


Figure 12
The compression resistance of the hardened pellets

6 Analysis and Discussion

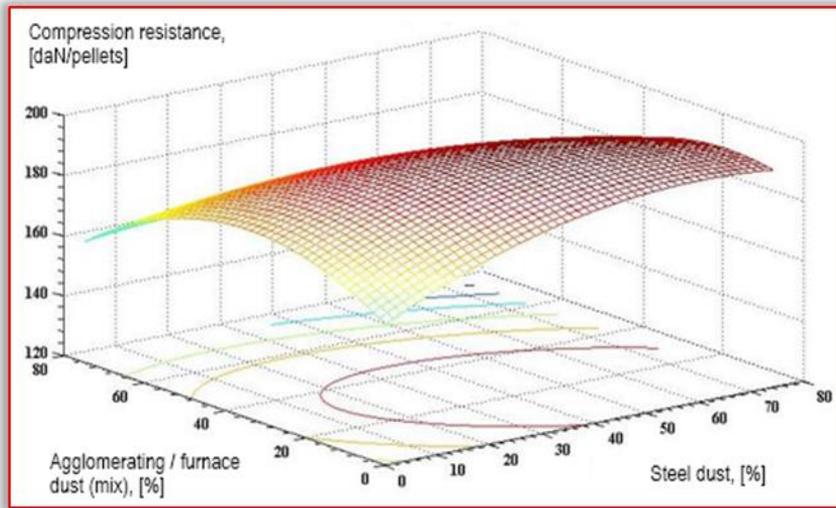
Our analysis of the study of the influence of solid components with iron content (steel dust, agglomeration–furnace dust, red mud/bauxide residue and anti-corrosive sludge) on compression resistance, are graphically represented in Figures 13-17, based on the results obtained in our experiments.

In diagram presented in Figure 13, [$R_c = f(\text{steel dust, agglomeration–furnace dust})$], it is noted that there is an area with a maximum point, and the range with values greater than 180 daN/pellet is quite extensive, so that the variation limits for the two components are quite high (approximately 60 agglomeration–furnace dust, over 70% steel dust). It is noted that only at low values of the two components (approximately 60% agglomeration–furnace dust, or over 70% steel dust), the resistance to compression is less than 180 daN/pellet. However, at values of only 40% agglomeration–furnace dust and between 30-60% steel dust, compression resistance reaches values of almost 200 daN/pellet.

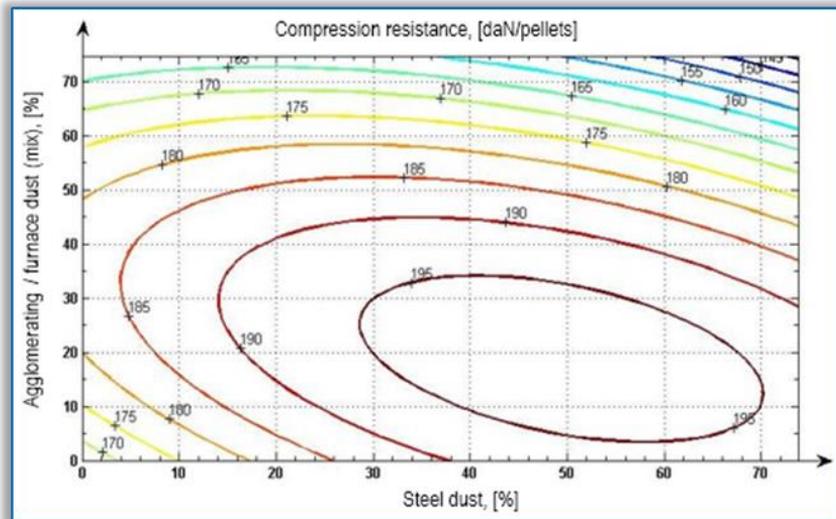
In the diagram presented in Figure 14, [$R_c = f(\text{steel dust, bauxide residue})$], it is found that there is an increase in compression resistance with an increase in the proportion of red mud/bauxide residue and steel dust, the best values being obtained for the upper limits of the two components of the recipe. The technological explanation lies in the high finesse of both materials and the content of Al_2O_3 in bauxide residue. Depending on the availability of waste assortments it would be preferable to place their proportion at quantities that provide compression resistance above 200 daN/pellet – in the right corner, top of the diagram (10-14% bauxide residue and 30-70% steel dust) or more than 180-190 daN/pellet (10-14% bauxide residue and between 10-70% steel dust).

The diagram presented in Figure 15 [$R_c = f(\text{agglomeration–furnace dust, bauxide residue})$] shows the positive influence of bauxide residue, practically an addition of this between 4-16% ensures good values for the compression resistance of the pellets, even if the content of agglomeration–furnace dust does not exceed 50%. Good values of compression resistance are also obtained for the level curves corresponding to some participations in recipes with 10-50% agglomeration powder–furnaces and smaller amounts of bauxide residue (2-6%).

In the case of use of anti–corrosive sludge together with agglomeration–furnace dust – Figure 16 [$R_c = f(\text{agglomeration–furnace dust, anti–corrosive sludge})$] –, it is found that higher values for compression resistance are obtained at their lower proportions in the pelletization charge. Very good quality pellets can be obtained with agglomeration–furnace dust within a maximum of 50% and for 10-15% anti-corrosive sludge, or even lower (5-10%).



(a) the regression surface described by the laboratory data



(b) the level curves of compression resistance, in 2D coordinates

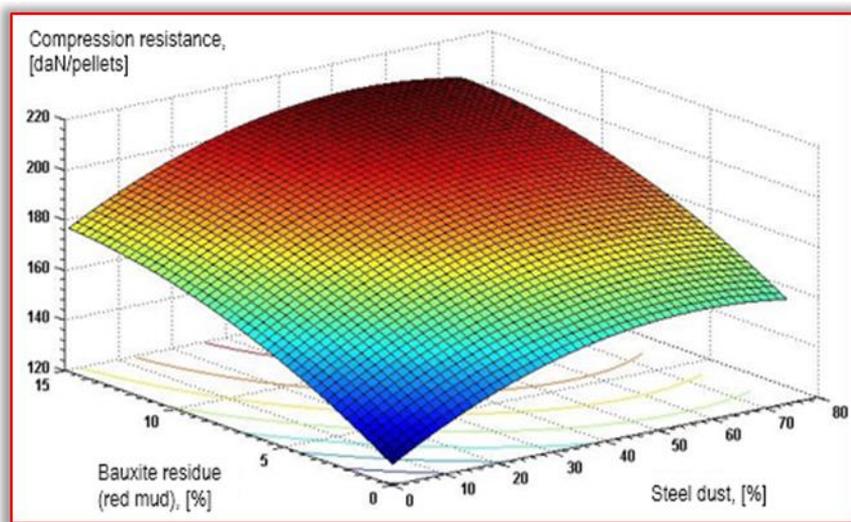
Figure 13

Compression resistance of hardened pellet's with the proportions of agglomeration–furnace dust and steel dust used in the experimental recipes – [Rc = f(agglomeration–furnace dust, steel dust)]

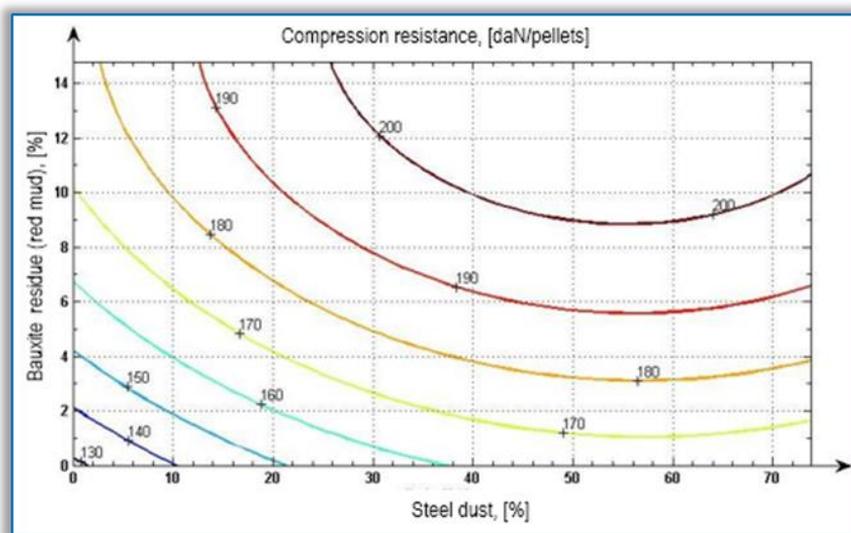
Equation type: $z = a_{(1)}x^2 + a_{(2)}y^2 + a_{(3)}xy + a_{(4)}x + a_{(5)}y + a_{(6)}$, in which the coefficients are:

$a_{(1)} = -0.0074$; $a_{(2)} = -0.0137$; $a_{(3)} = -0.0085$; $a_{(4)} = 0.8945$; $a_{(5)} = 0.9330$; $a_{(6)} = 166.7903$

Coefficient of multiple correlation: $R^2 = 0.7482$ (relative high grade of correlation)



(a) the regression surface described by the laboratory data



(b) the level curves of compression resistance, in 2D coordinates

Figure 14

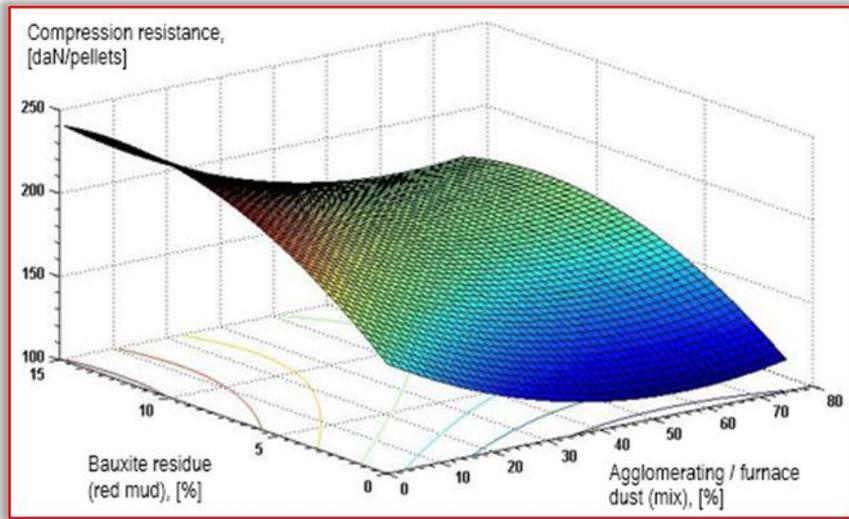
Compression resistance of hardened pellet's with the proportions of steel dust and bauxite residue / red

mud used in the experimental recipes – $[R_c = f(\text{steel dust, bauxite residue / red mud})]$

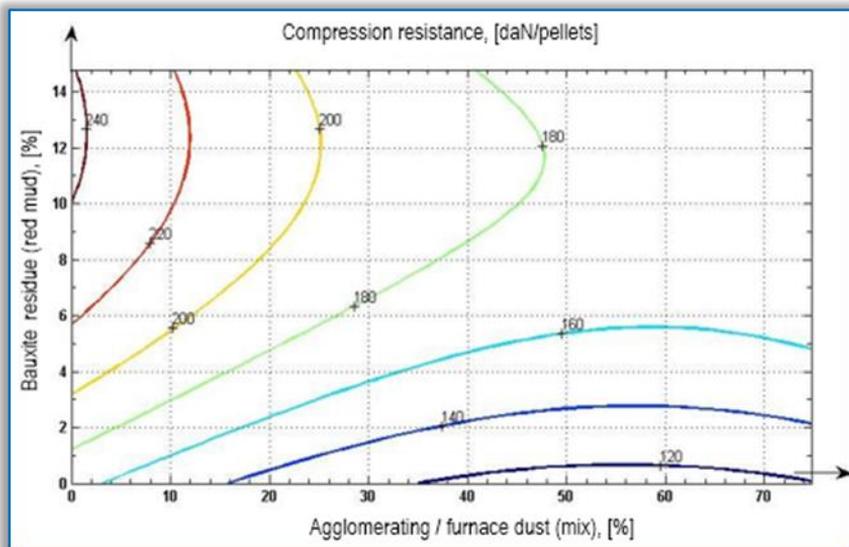
Equation type: $z = a_{(1)}x^2 + a_{(2)}y^2 + a_{(3)}xy + a_{(4)}x + a_{(5)}y + a_{(6)}$, in which the coefficients are:

$a_{(1)} = -0.0109$; $a_{(2)} = -0.1750$; $a_{(3)} = -0.0055$; $a_{(4)} = 1.2592$; $a_{(5)} = 5.8883$; $a_{(6)} = 128.1288$

Coefficient of multiple correlation: $R^2 = 0.9389$ (high grade of correlation)



(a) the regression surface described by the laboratory data



(b) the level curves of compression resistance, in 2D coordinates

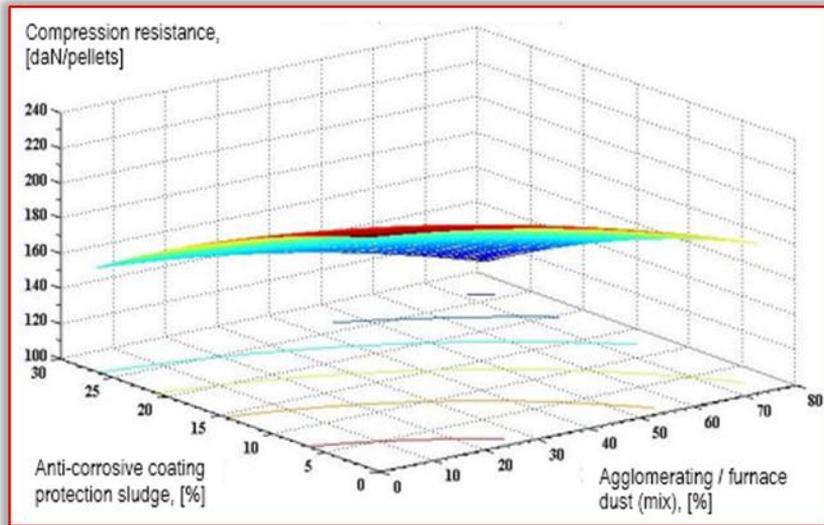
Figure 15

Compression resistance of hardened pellet's with the proportions of agglomeration–furnace dust and bauxite residue used in recipes – $[Rc = f(\text{agglomeration–furnace dust, bauxite residue})]$

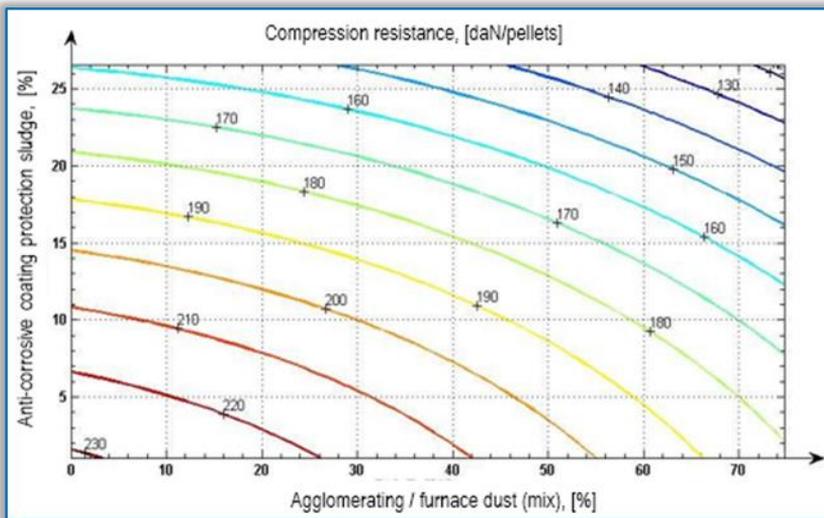
Equation type: $z = a_{(1)}x^2 + a_{(2)}y^2 + a_{(3)}xy + a_{(4)}x + a_{(5)}y + a_{(6)}$, in which the coefficients are:

$a_{(1)} = 0.0173$; $a_{(2)} = -0.4860$; $a_{(3)} = -0.0193$; $a_{(4)} = -1.9154$; $a_{(5)} = 12.2658$; $a_{(6)} = 165.8485$

Coefficient of multiple correlation: $R^2 = 0.8652$ (high grade of correlation)



(a) the regression surface described by the laboratory data



(b) the level curves of compression resistance, in 2D coordinates

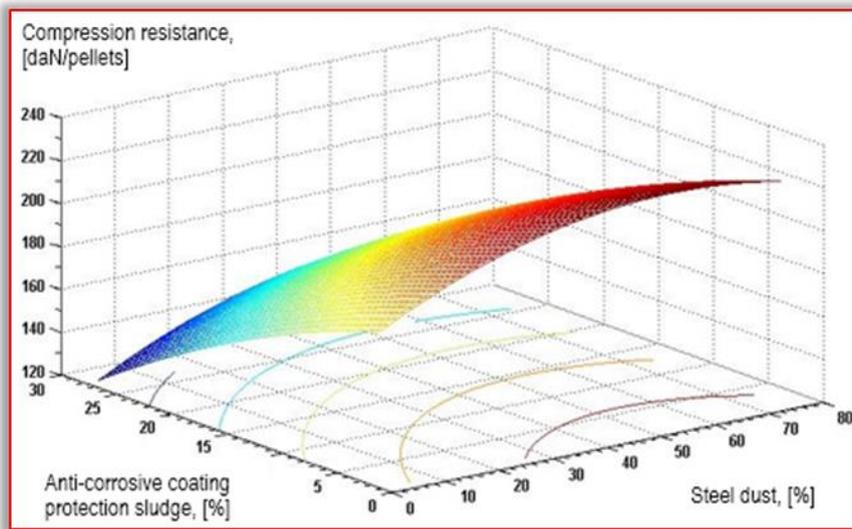
Figure 16

Compression resistance of hardened pellets with the proportions of agglomeration–furnace dust and anti–corrosive sludge used in recipes – $[Rc=f(\text{agglomeration–furnace dust, anti–corrosive sludge})]$

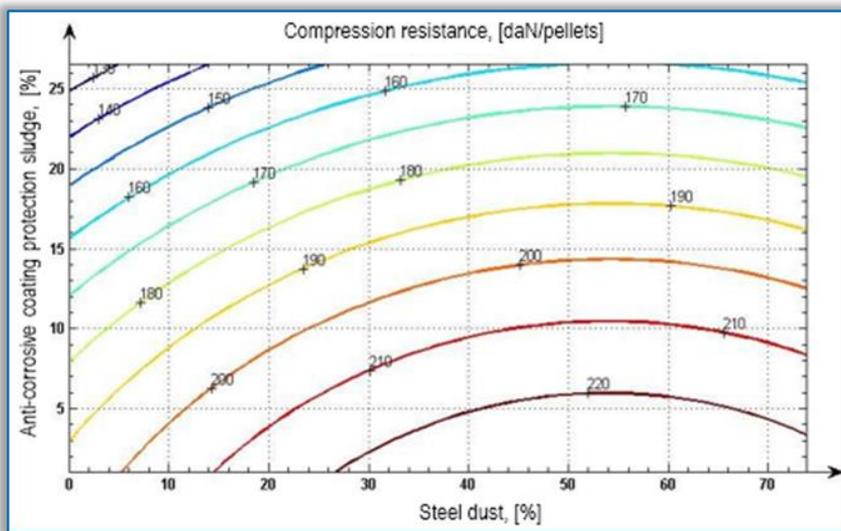
Equation type: $z = a_{(1)}x^2 + a_{(2)}y^2 + a_{(3)}xy + a_{(4)}x + a_{(5)}y + a_{(6)}$, in which the coefficients are:

$a_{(1)} = -0.0050$; $a_{(2)} = -0.0430$; $a_{(3)} = 0.0037$; $a_{(4)} = -0.2944$; $a_{(5)} = -1.6302$; $a_{(6)} = 232.8057$

Coefficient of multiple correlation: $R^2 = 0.8742$ (high grade of correlation)



(a) the regression surface described by the laboratory data



(b) the level curves of compression resistance, in 2D coordinates

Figure 17

Compression resistance of hardened pellet's with the proportions of steel dust and anti-corrosive sludge used in the recipes – $[Rc=f(\text{steel dust, anti-corrosive sludge})]$

Equation type: $z = a_{(1)}x^2 + a_{(2)}y^2 + a_{(3)}xy + a_{(4)}x + a_{(5)}y + a_{(6)}$, in which the coefficients are:

$a_{(1)} = -0.0124$; $a_{(2)} = -0.0420$; $a_{(3)} = 0.0007$; $a_{(4)} = 1.3304$; $a_{(5)} = -1.5737$; $a_{(6)} = 194.9876$

Coefficient of multiple correlation: $R^2 = 0.8967$ (high grade of correlation)

The increase in the proportion of steel dust to 30-70% of the pelletizing charge has a positive influence on the resistance to compression of the pellets, but the anticorrosive sludge have a negative influence at increases in quantities, preferably a use of only 5-10%, as is presented in the Figure 17 [$R_c = f(\text{steel dust, anti-corrosive sludge})$].

Our analysis on the compression resistance of the pellets provided by the additions (bentonite, lime and graphite) are graphically represented in Figures 18-20, based on the results obtained in our experiments. Also, we shows the lime dependence with the two main components (steel dust and agglomeration–furnace dust), in Figure 21 [$R_c = f(\text{steel dust, lime})$] and Figure 22 [$R_c = f(\text{agglomeration–furnace dust, lime})$]. With regard to these cases of analysis, the following technological remarks are required:

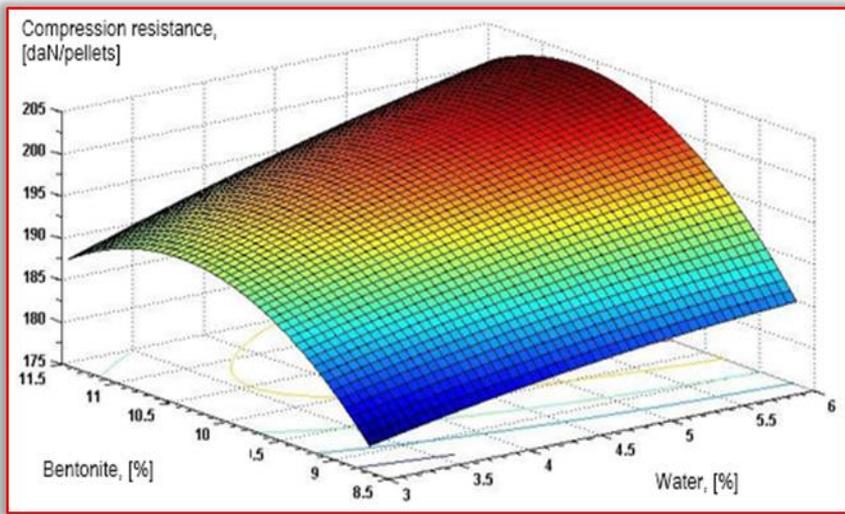
- Increasing the proportion of water to 4-6% and bentonite to about 9% leads to an increase in compression resistance, as appropriate conditions have been provided to obtain “green” pellets with high strength, from powdery materials, which is also found in the burned pellets (Figure 18 [$R_c = f(\text{water, bentonite})$]).
- The proportions of the water and lime additions of are well correlated in most cases, with the exception of the addition of lime below 1% and water below 4%, respectively the addition of lime above 5-6%, where the compression resistance is below the required limit. The field of work is very extensive, virtually unlimited, and from this point of view there are no particular problems (Figure 19 [$R_c = f(\text{water, lime})$]).

The proportions of lime and bentonite are well chosen, so that they have close influence and can substitute each other, ensuring for compression resistance values >180 daN/pellet (Figure 20 [$R_c = f(\text{lime, bentonite})$]).

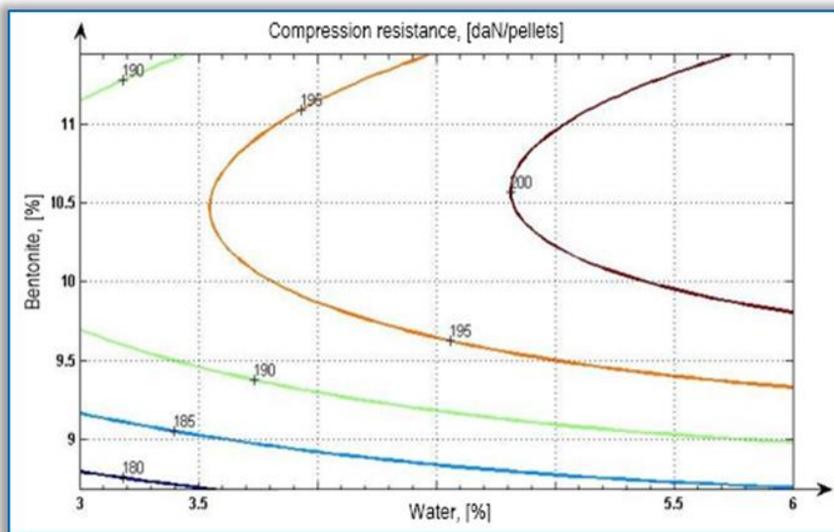
The addition of graphite is insignificant, and the quantity used in recipes, does not influence the process of pelletization nor the quality of the burned pellet. For this reason, graphic dependencies are not shown among these results.

Increased resistance to compression of the pellets, slightly decreases with the increase in the proportion of agglomeration–furnace dust and instead increases with the addition of lime, according to the Figure 21 [$R_c = f(\text{agglomeration–furnace dust, lime})$]. The agglomeration–furnace dust also has a positive effect, in that, in the combustion process, it ensures the reduction of iron oxides, increasing the degree of metalization of the pellets.

Finally, the chart presented in Figure 22 [$R_c = f(\text{steel dust, lime})$] shows that, in terms of the correlation of steel dust and the addition of lime, practically regardless of their proportion and the limits of the addition of lime, up to 6%, consistent pellets are obtained for use in steelmaking aggregates.



(a) the regression surface described by the laboratory data



(b) the level curves of compression resistance, in 2D coordinates

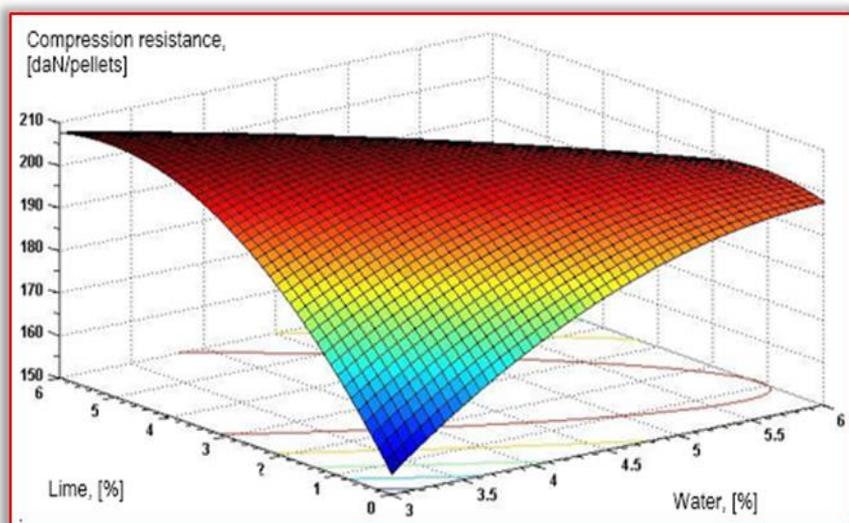
Figure 18

Compression resistance of hardened pellet's with the proportions of bentonite and water used in the experimental recipes – [Rc = f(bentonite, water)]

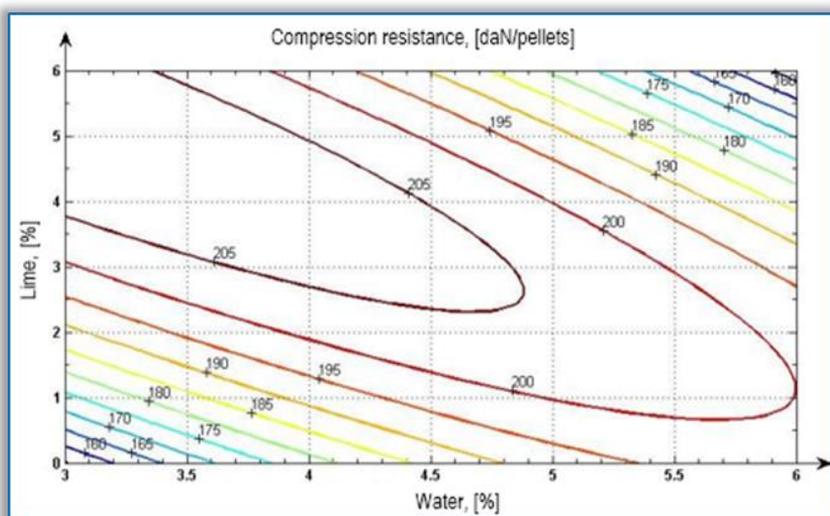
Equation type: $z = a_{(1)}x^2 + a_{(2)}y^2 + a_{(3)}xy + a_{(4)}x + a_{(5)}y + a_{(6)}$, in which the coefficients are:

$a_{(1)} = -0.4072$; $a_{(2)} = -1.1150$; $a_{(3)} = -4.7305$; $a_{(4)} = 0.8048$; $a_{(5)} = 96.2018$; $a_{(6)} = -314.4748$

Coefficient of multiple correlation: $R^2 = 0.7922$ (high grade of correlation)



(a) the regression surface described by the laboratory data



(b) the level curves of compression resistance, in 2D coordinates

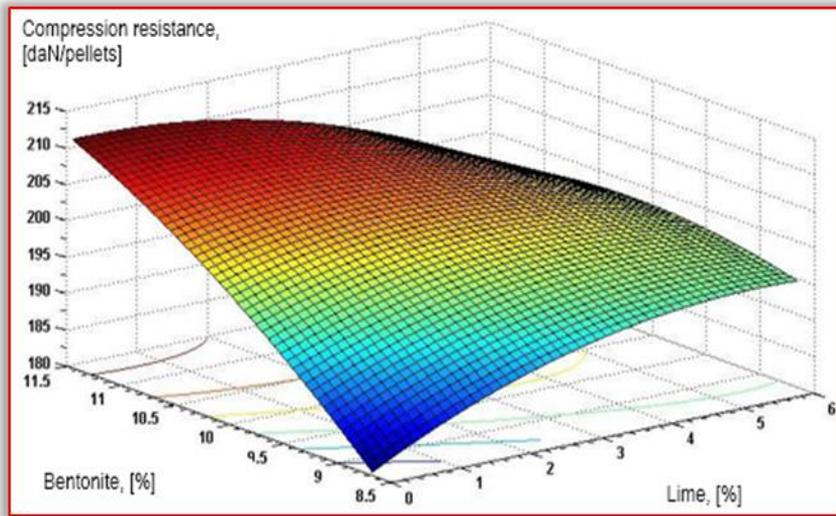
Figure 19

Compression resistance of hardened pellet's with the proportions of lime and water used in the experimental recipes – [Rc = f(lime, water)]

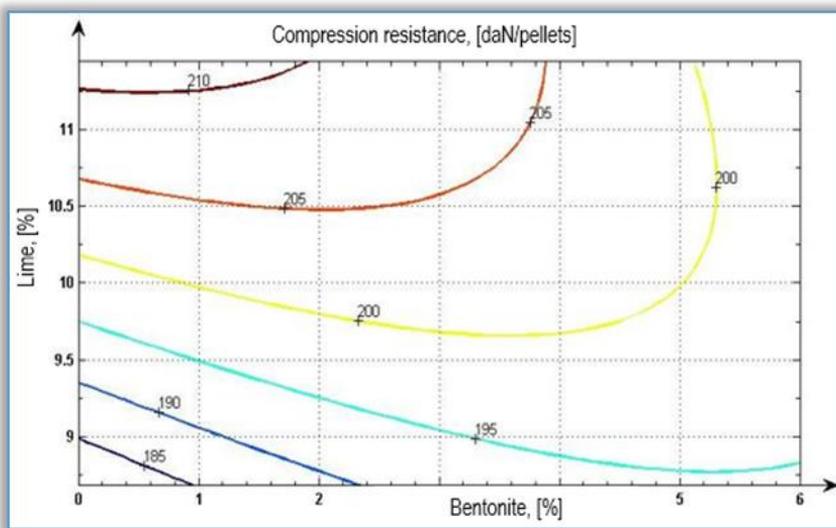
Equation type: $z = a_{(1)}x^2 + a_{(2)}y^2 + a_{(3)}xy + a_{(4)}x + a_{(5)}y + a_{(6)}$, in which the coefficients are:

$a_{(1)} = -4.4814$; $a_{(2)} = -2.0376$; $a_{(3)} = -5.4652$; $a_{(4)} = 54.5467$; $a_{(5)} = 37.3916$; $a_{(6)} = 31.4430$

Coefficient of multiple correlation: $R^2 = 0.6056$ (relative high grade of correlation)



(a) the regression surface described by the laboratory data



(b) the level curves of compression resistance, in 2D coordinates

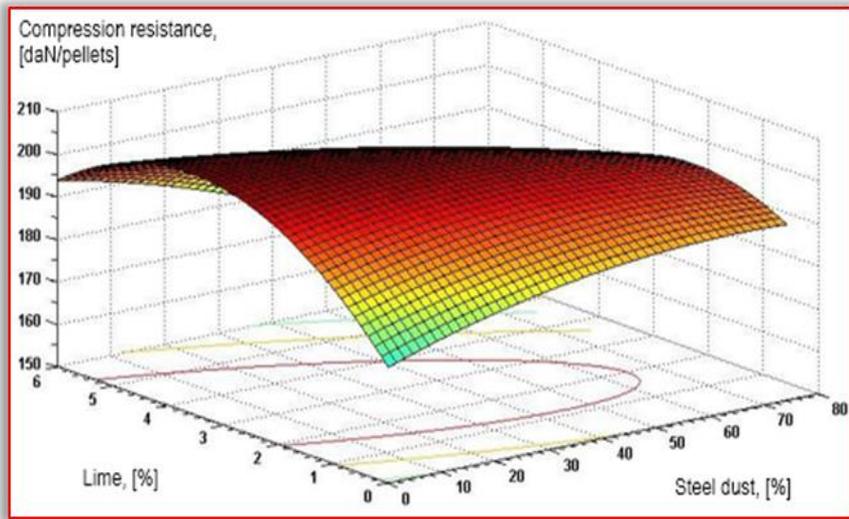
Figure 20

Compression resistance of hardened pellet's with the proportions of lime and bentonite used in the experimental recipes – [Rc = f(lime, bentonite)]

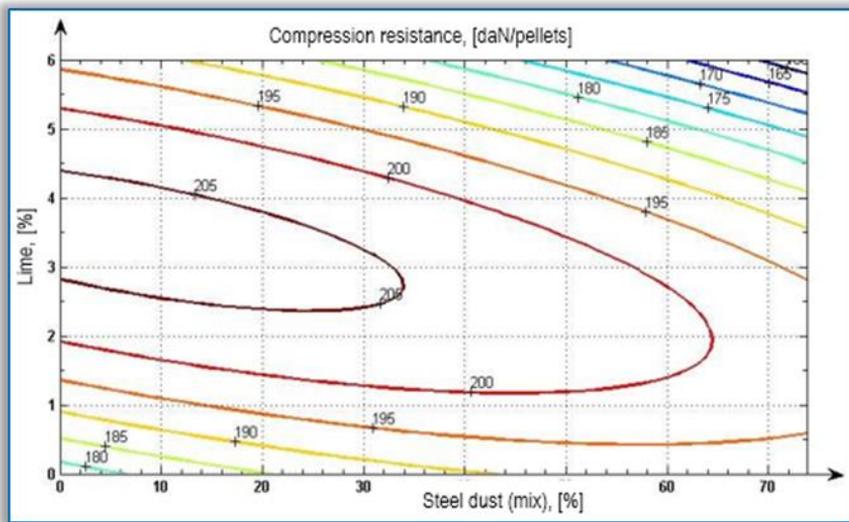
Equation type: $z = a_{(1)}x^2 + a_{(2)}y^2 + a_{(3)}xy + a_{(4)}x + a_{(5)}y + a_{(6)}$, in which the coefficients are:

$a_{(1)} = -0.4720$; $a_{(2)} = -1.4200$; $a_{(3)} = -1.7941$; $a_{(4)} = 20.7279$; $a_{(5)} = 39.7588$; $a_{(6)} = -57.6807$

Coefficient of multiple correlation: $R^2 = 0.8698$ (high grade of correlation)



(a) the regression surface described by the laboratory data



(b) the level curves of compression resistance, in 2D coordinates

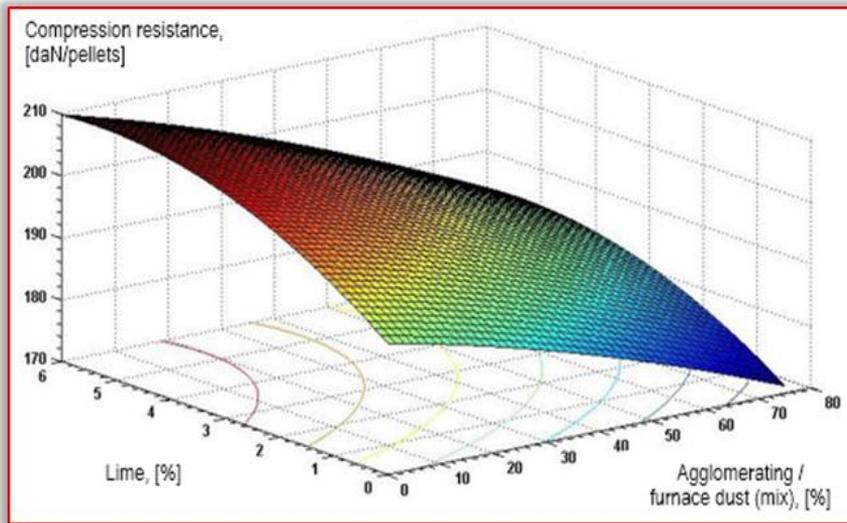
Figure 21

Compression resistance of hardened pellet's with the proportions of steel dust and lime used in the experimental recipes – [Rc = f(steel dust, lime)]

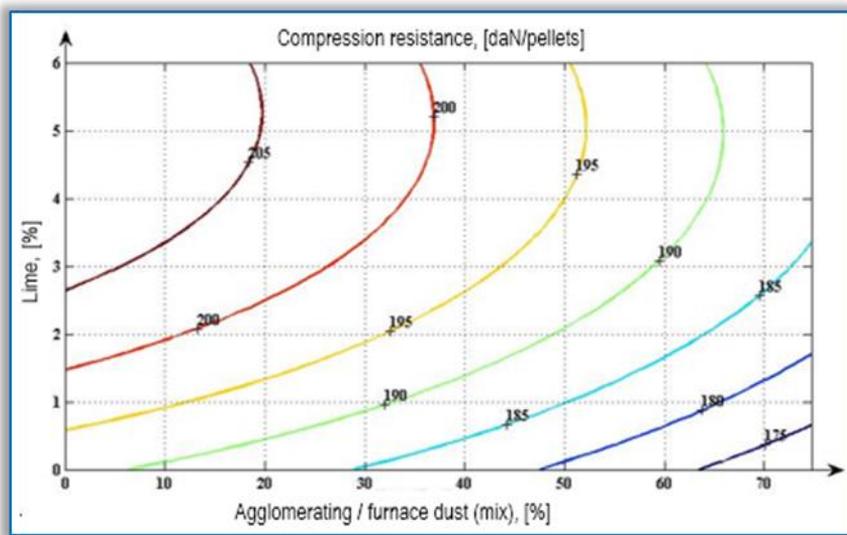
Equation type: $z = a_{(1)}x^2 + a_{(2)}y^2 + a_{(3)}xy + a_{(4)}x + a_{(5)}y + a_{(6)}$, in which the coefficients are:

$a_{(1)} = -0.0012$; $a_{(2)} = -0.6413$; $a_{(3)} = -0.0085$; $a_{(4)} = -0.1783$; $a_{(5)} = 6.9270$; $a_{(6)} = 191.1665$

Coefficient of multiple correlation: $R^2 = 0.7759$ (high grade of correlation)



(a) the regression surface described by the laboratory data



(b) the level curves of compression resistance, in 2D coordinates

Figure 22

Compression resistance of hardened pellet's with the proportions of agglomeration–furnace dust and lime used in the experimental recipes – [Rc = f(agglomeration–furnace dust, lime)]

Equation type: $z = a_{(1)}x^2 + a_{(2)}y^2 + a_{(3)}xy + a_{(4)}x + a_{(5)}y + a_{(6)}$, in which the coefficients are:

$a_{(1)} = -0.0034$; $a_{(2)} = -2.2408$; $a_{(3)} = -0.1161$; $a_{(4)} = 0.4437$; $a_{(5)} = 16.1699$; $a_{(6)} = 177.2020$

Coefficient of multiple correlation: $R^2 = 0.8171$ (high grade of correlation)

Conclusions

In this paper the research and relevant results are presented, in regard to the formation of pellets, using waste from ferrous industry (steel dust, agglomerating–furnace dust) and mining and mineral processing sectors (red mud/bauxide residue, anti-corrosive sludge). For this, we considered the existing ferrous powdery waste in the Hunedoara County (Hunedoara and Calan) area and several powdery wastes from Bihor County (Oradea) area. In addition, within the pellets recipes, graphite is used, as the reducing agent and respectively, bentonite and lime are used as binders.

Our research work has considered the following technological problems:

- The wetting capacity of processed materials, in the pelletizer scale
- The quality of “green” pellets according to the processed materials, their chemical composition and granulation, the addition of water and binders
- The quality of the hardened pellets according to the processed solid materials, the chemical composition and their obtained dimensions, the addition of water and the binders quantity
- The compression resistance of the burned pellets according to the solid materials processed, the addition of water and binders

On the basis of the above, we studied the possibilities of recovery of powdery ferrous waste in the steel industry. Given that at the national level, this activity, unfortunately, has a very limited scope. Our strategies should focus on development of recycling/valorisation capacities of iron containing raw materials contained in waste, as well as on installations and technologies using waste in the production process. In this sense, analyzing the results obtained in our laboratory, it follows that various waste processed in the form of pellets, according to the experimental recipes #1-6, can be further utilized in industrial applications.

Acknowledgment

The study was carried out on the basis of extensive research over the last 20 years, undertaken within the Hunedoara Faculty of Engineering, following the results of numerous researches [7, 9, 10, 13-26] that concerned a large number of researchers in this institution. This research presented now is part of the research on the doctoral thesis [13], as well as a number of researches undertaken after this period in order to recover by pelletization of materials containing valuable or deficitary elements from different kinds of powdery waste stored in areas with risk factor.

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